



candy-making science

splash 2021: caramel edition



mise en place



Ingredients

Materials

Fats:

- Dairy: $\frac{1}{2}$ cup heavy cream and 2 tbsp butter
- Nondairy: $\frac{2}{3}$ cup full fat coconut milk

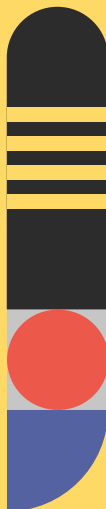
Sugar things:

- $\frac{3}{4}$ cup sugar
- 2 tbsp corn syrup
- 2 tbsp water

Other:

- If using unsalted butter or coconut milk, a little bit of salt
- Flavoring, like vanilla (optional)

- 2 Saucepans, one of them at least medium sized
- Measuring cups/spoons
 - ($\frac{1}{2}$ cups, $\frac{1}{3}$ cups, $\frac{1}{4}$ cups, tablespoon)
- Greased up heatsafe flat surface (cookie tray, baking sheet, largish tray, etc)
- Cooking Spoon
- [Cup with cold water and a spoon] or [a Candy thermometer]
- Whisk (Optional)



Reminder:
Dairy: $\frac{1}{2}$ cup heavy cream and 2 tbsp butter
Nondairy: $\frac{2}{3}$ cup full fat coconut milk

let's get started: warm the fats

- Put your fat(s) into the smaller of the two saucepans
 - A pinch of salt if using unsalted butter or coconut milk
- Place over medium-low heat for about 5 minutes
 - Around 4-ish on a 1-10 numbered stove
- Goals
 - Dairy: a relatively homogeneous liquid with the cream and butter mix, to a yellow-ish color
 - Nondairy: heat it up until it is all liquid

Reminder:
¾ cup sugar, 2 tbsp corn syrup,
2 tbsp water

in the meantime... sugar paste

● **NOT OVER HEAT**

- Put your sugar things into the larger of the two saucepans
- Mix it all into a thick grainy paste
- Attach your candy thermometer now, if using one





back to the fats: homogeneous mixture!

~5 minutes in

Notice that it is a yellow-ish color. Some yellow spots from the butter are ok, just make sure it is mostly well mixed in

For those of you using coconut milk, it should just be coconut milk that's all liquid (the fat dissolved ☐).

Now, remove this from heat and put it somewhere off to the side where it won't get cold too fast



SUGAR TIME

**Let's put the sugar paste
on medium heat**

Around 5-ish on a 1-10 numbered stove

**Don't mix it!
Let it come to a boil**



the cold water test (wow)

- Using your cooking spoon, take a bit of the mix, and drop it into the cup of water
- At this point, the resulting ball should fill fairly squishy (fish it out with a spoon)

~3 minutes in





Soft Ball

230°F- 240°F

squishy ball, kinda
feels like play-doh
tbh

Firm Ball

240°F- 250°F

noticeably
tougher, feels like
modeling clay

Hard Ball

250°F- 265°F

feels like a raw
almond (very
tough)

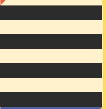
Soft Crack

265°F- 290°F

makes a soft cracking
noise when dropped,
forms pliable threads

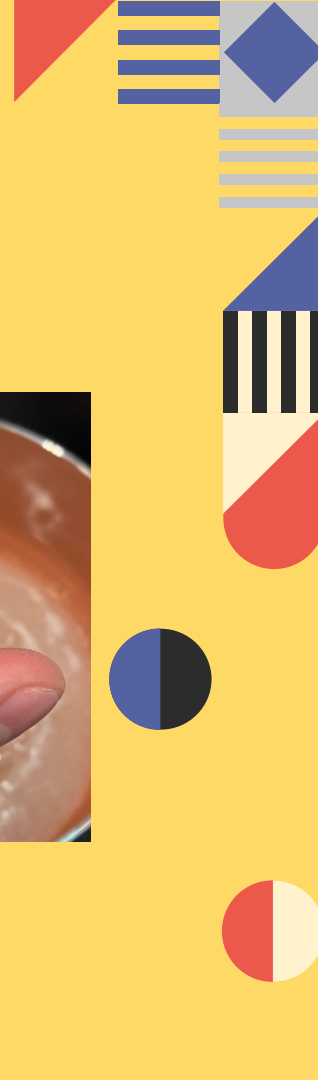
THIS IS WHAT WE WANT

**and now,
a montage
of various
images**



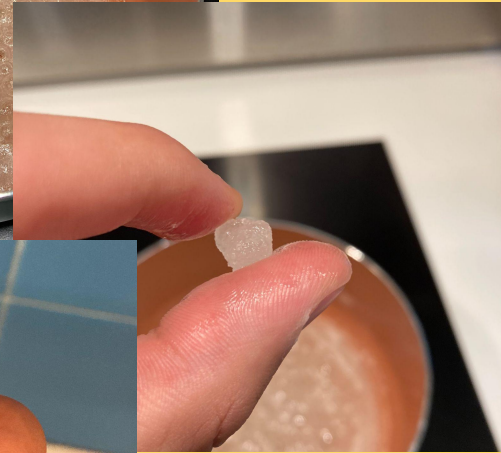
firm ball

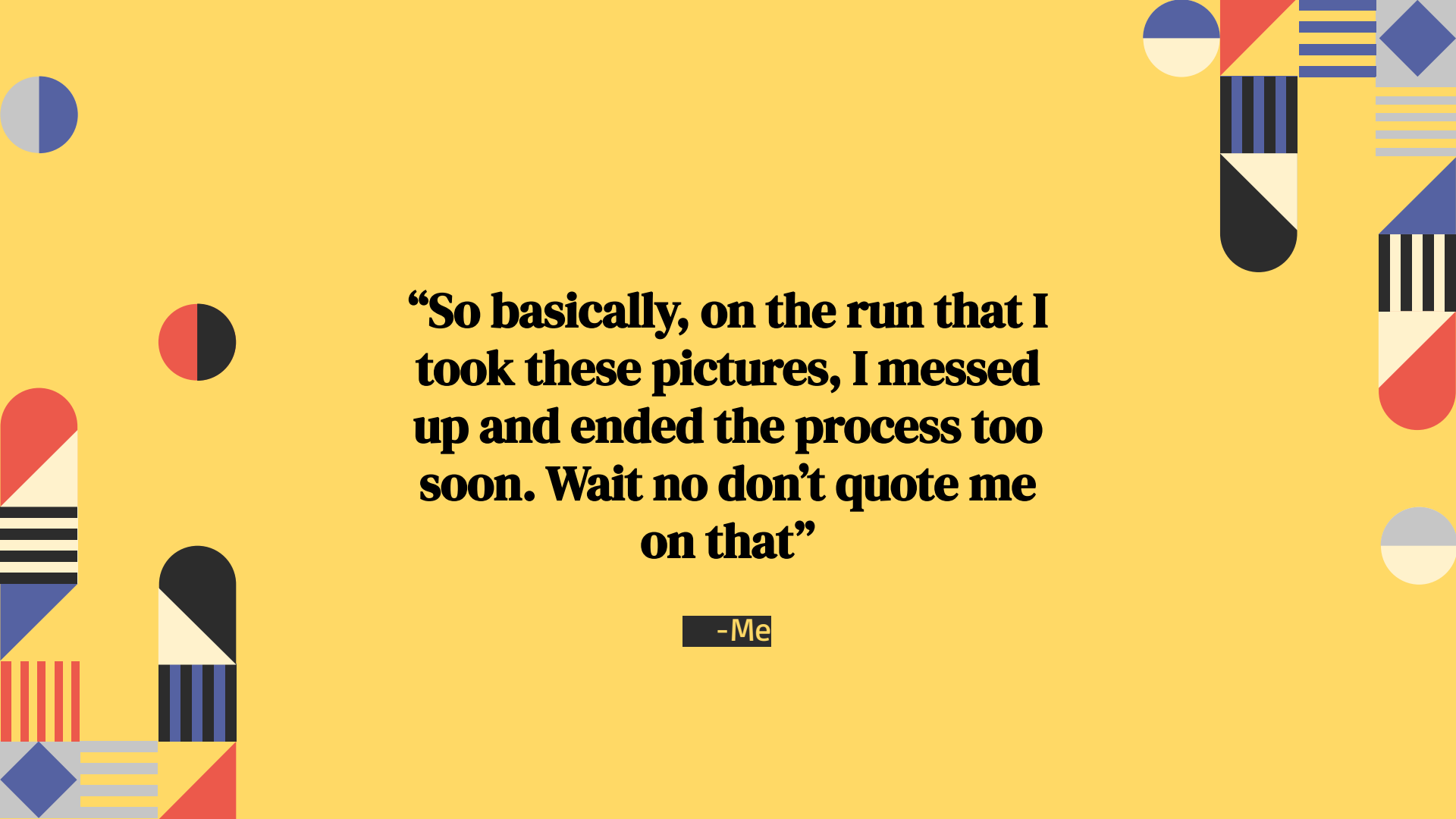
240°F- 250°F
~4 minutes in



hard-ish ball

250°F- 265°F
~6 minutes in





“So basically, on the run that I took these pictures, I messed up and ended the process too soon. Wait no don’t quote me on that”

-Me

beginning soft crack

265°F- 280°F
~7 minutes in



image not available

lol oops

PAUSE

pouring in the fats

- **TAKING IT OFF THE HEAT**

- slowly pour in the fats, whisking them in as you pour
- if you don't have a whisk, use the cooking spoon to mix it in as you pour



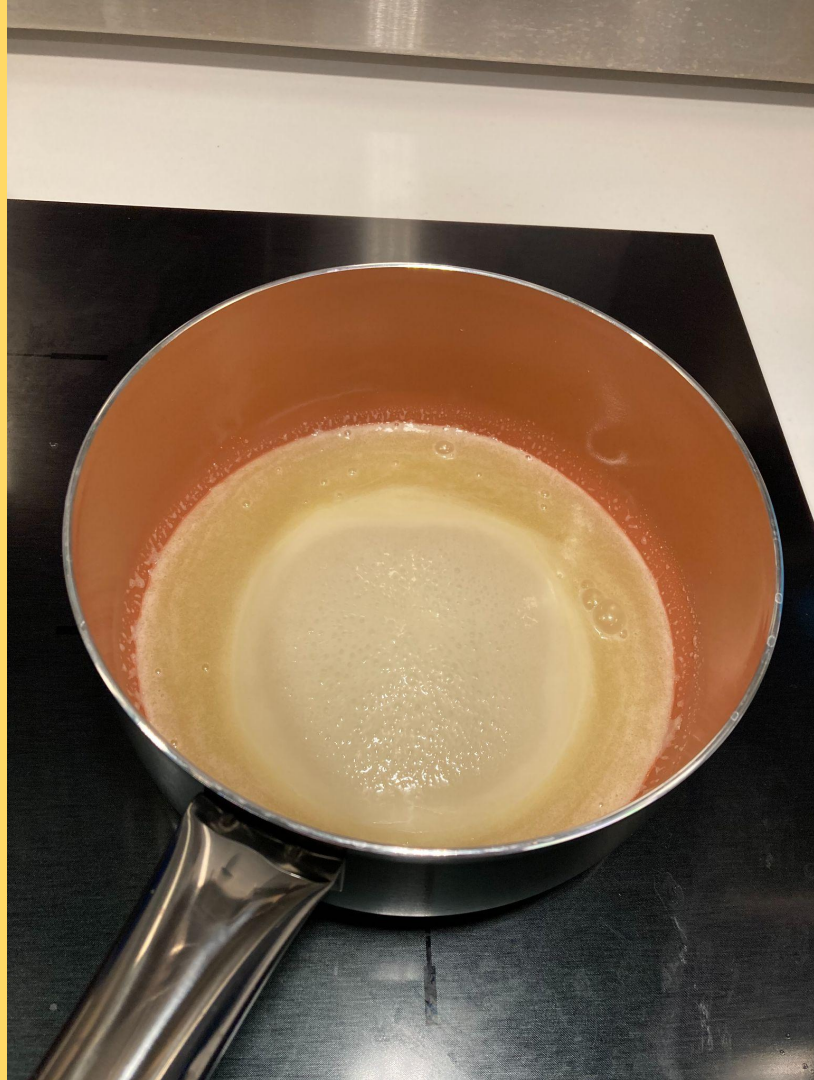
sugar-fat mixture

- what we just did is bring the temperature of the sugar mixture down by about 40°F, down to around ~230°F
- let's put it back on medium heat
- separation here is fine, just let it do its thing



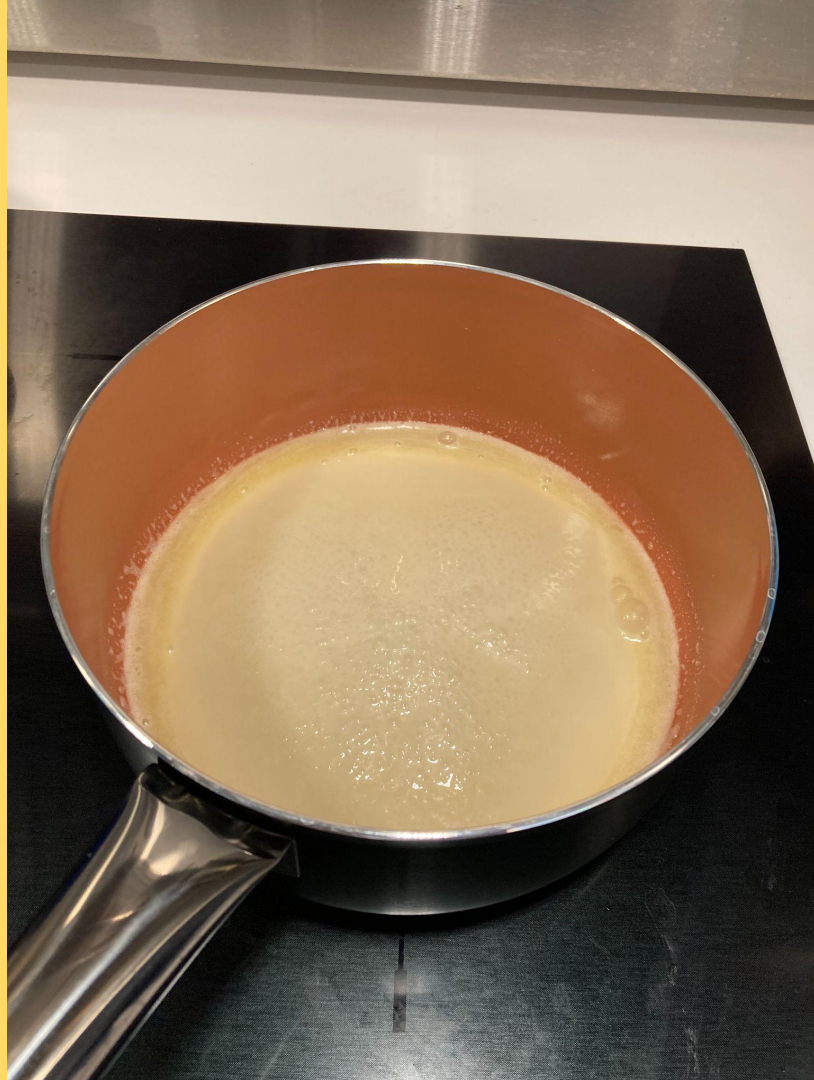
bubble time

230°F- 245°F
for about the next 4
minutes



bubble time

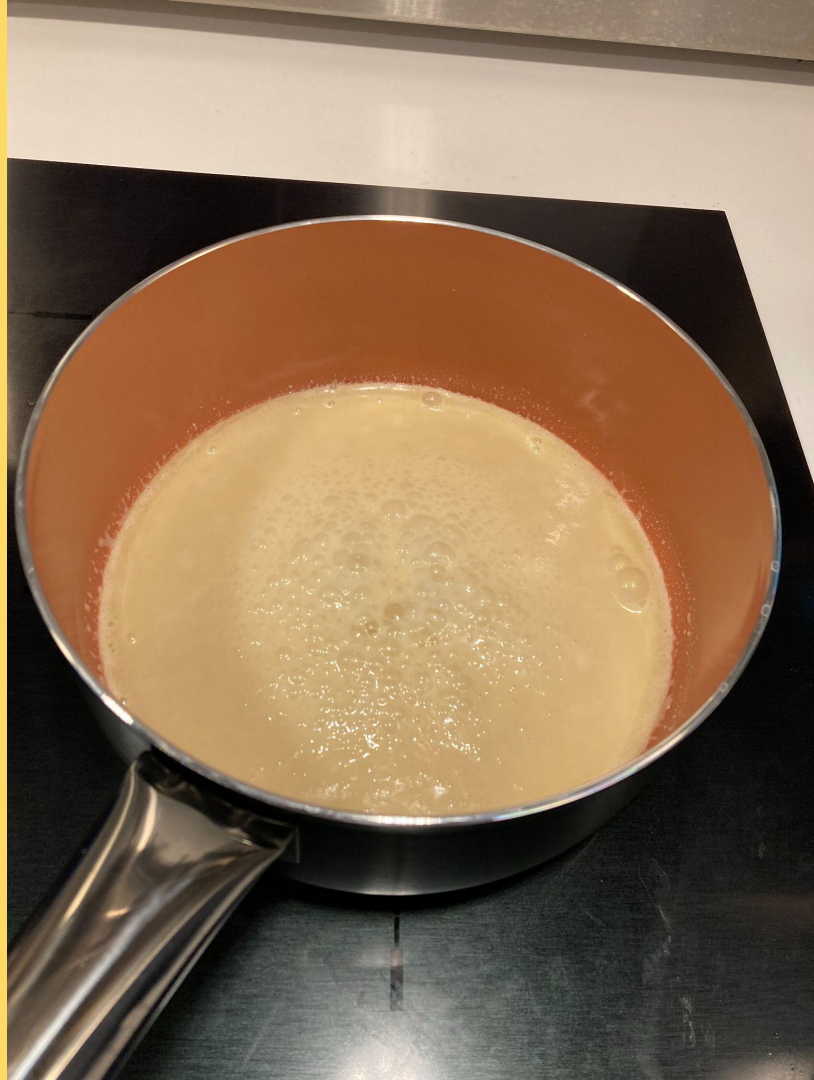
230°F- 245°F
for about the next 4
minutes



bubble time

230°F- 245°F

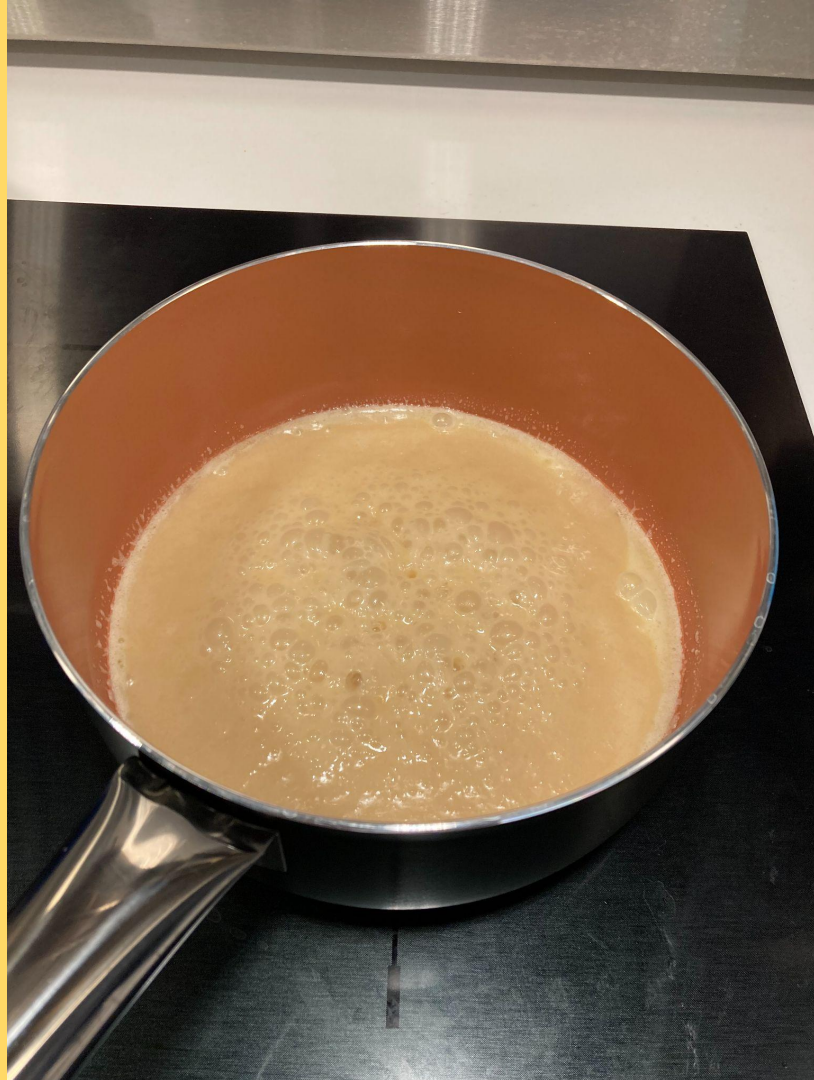
for about the next 4
minutes



bubble time

230°F- 245°F

for about the next 4
minutes



bubble time

230°F- 245°F
for about the next 4
minutes



bubble time

230°F- 245°F
for about the next 4
minutes



bubble time

230°F- 245°F

for about the next 4
minutes



almost done...

You can remove this from
heat anywhere between 245°F-250°
F, until you get the color you want

You're looking for a nice orange-ish or
brown-ish color at this point.

If you're adding a flavor,
remove from heat here, and quickly
mix in the flavor



image not available

i'm sorry i didn't get
a picture for this one: just go
with the color you
think looks nice

pouring time

pouring on our prepared greased
tray



all done

all done

(just let it cool for at least three hours)



the fats and the sugar stuff

- What is going on with these ingredients?
 - We broke down the ingredients into 2 main sections: the fats section and the sugar stuff section
 - The fats help prevent sugar from crystalizing in the candy
 - The sugars are there to actually make the candy lol
- What's the point of the two separated sections?

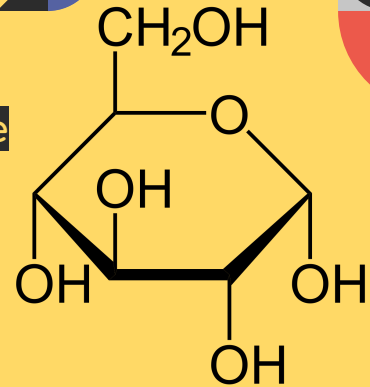
a closer look at sugar

So basically, there are three main basic sugars: glucose, fructose, and galactose. They are all monosaccharides (ie, they are the monomers of carbohydrates).

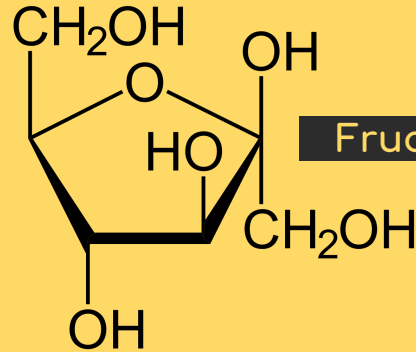
They all have the formula $C_6H_{12}O_6$.

Even though they have the same formula, they're put together in different ways: they are all constitutional isomers of each other.

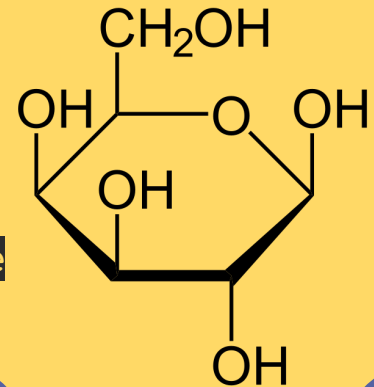
Glucose

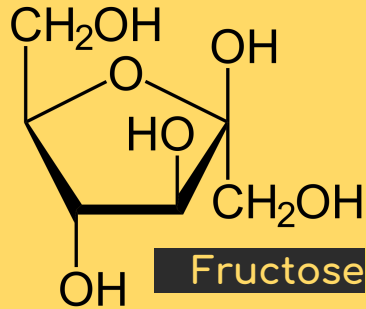
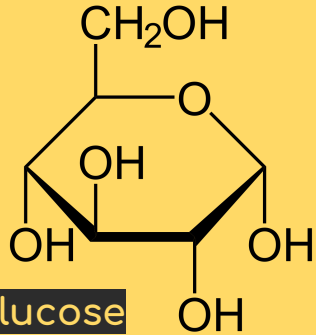
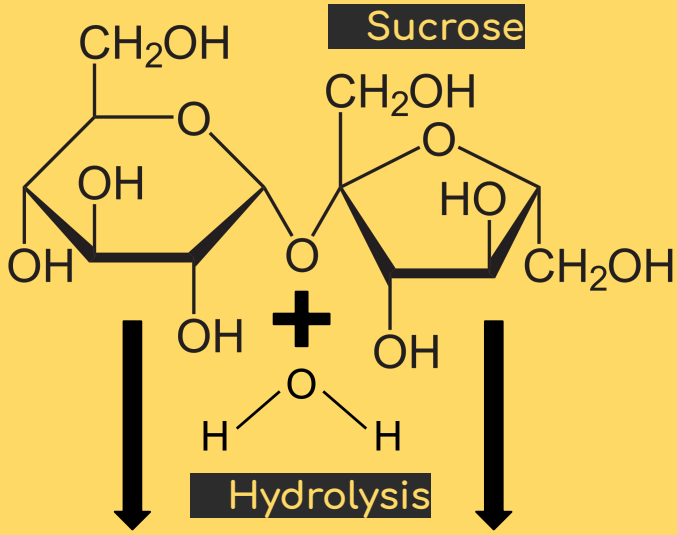


Fructose



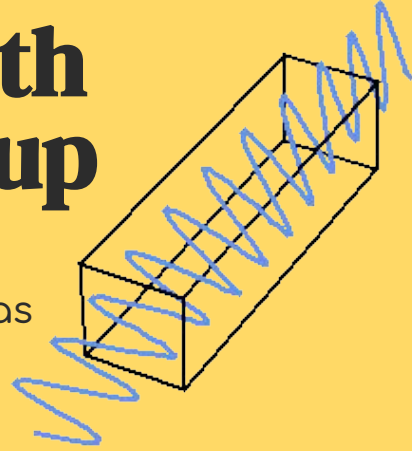
Galactose

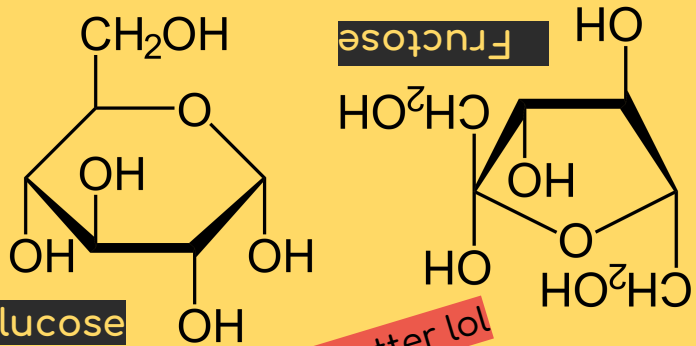
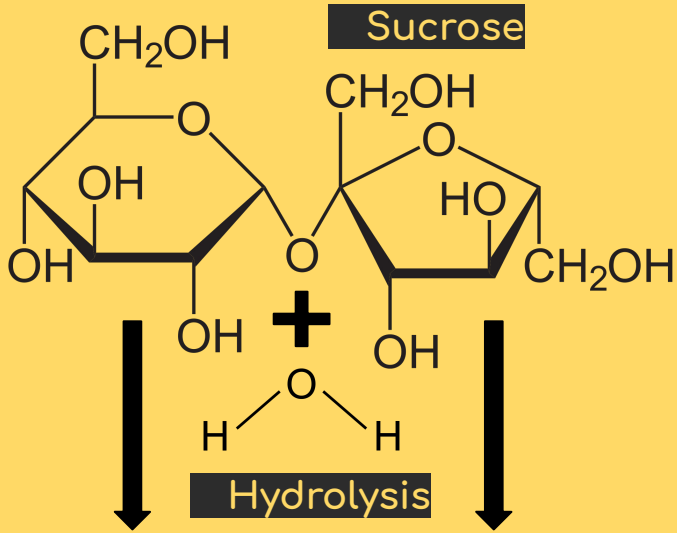




what's up with the corn syrup

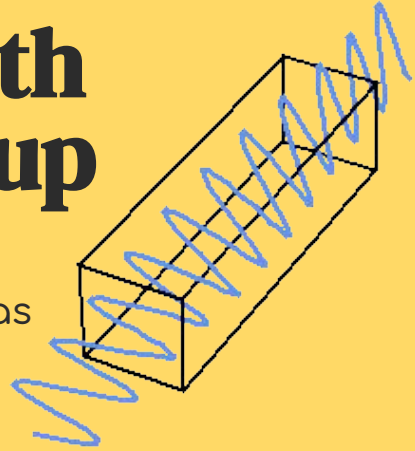
- Corn syrup is an invert sugar, meaning that it has the opposite optical rotation to that of its original sugar.
- These optical properties arise because sugars are chiral molecules.
- We make it by hydrolyzing sucrose (table sugar) into its monosaccharides, glucose and fructose.
- Why do we need it in the first place?





much better lol

what's up with the corn syrup

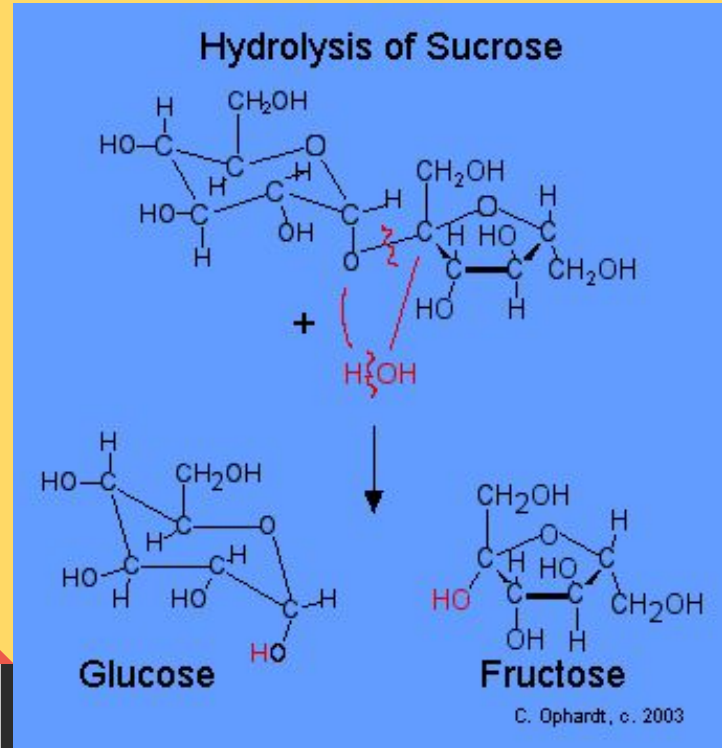


- Corn syrup is an invert sugar, meaning that it has the opposite optical rotation to that of its original sugar.
- These optical properties arise because sugars are chiral molecules.
- We make it by hydrolyzing sucrose (table sugar) into its monosaccharides, glucose and fructose.

Since corn syrup is just the monosaccharides, its molecules are small enough to get in the way of the sucrose (the rest of the sugar that we put in), which prevents it from crystallizing.

what's up with the corn syrup

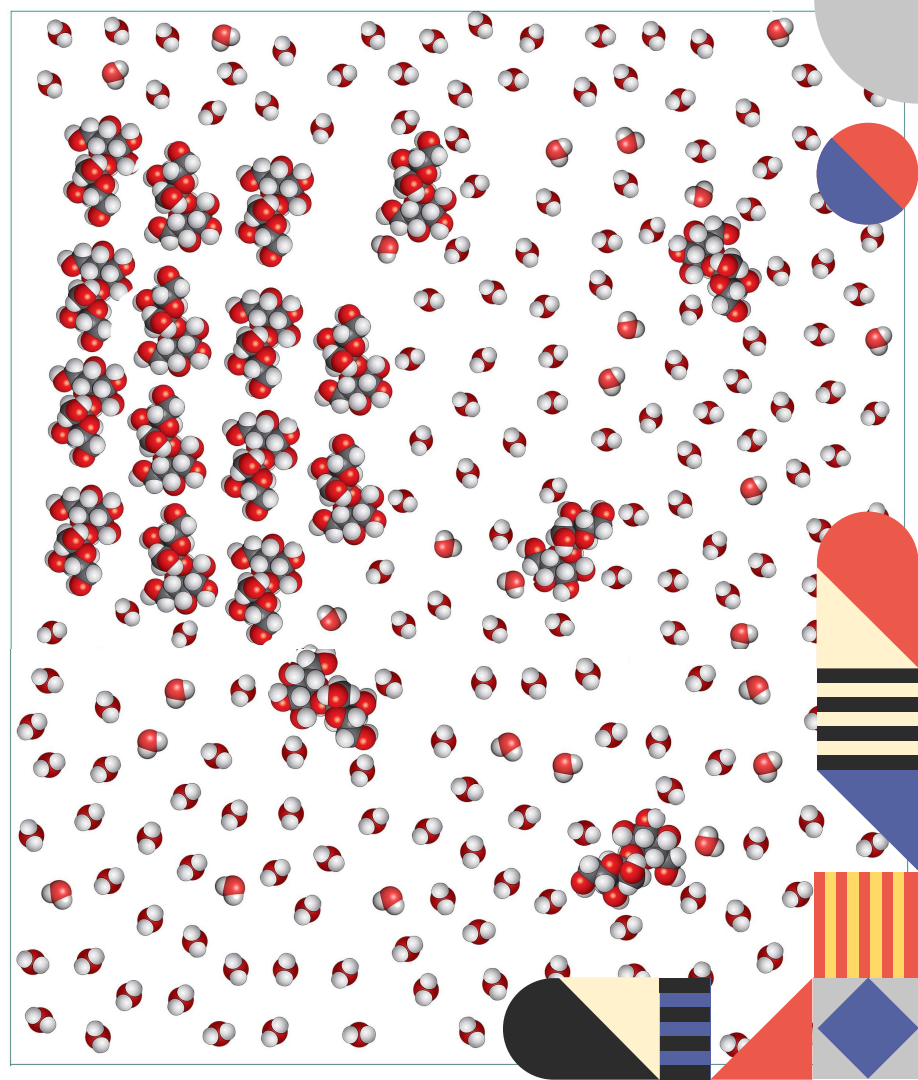
- As it turns out, hydrolyzing sucrose is fairly easy to do: all we need is table sugar (the sucrose), water, heat, and some acid to catalyze the reaction.
- You can make your own invert sugar using this reaction: the most common variation is using cream of tartar (an acid) or lemon juice as the catalytic base





why don't we want crystals?

- crystals forming in our candy means that table sugar (the sucrose) isn't fully dissolved
- if crystals form, then we get chunks and lumps of sugar (not great for smooth candies like caramels!)
- as mentioned before, the invert sugar and fats work together to prevent the sugar from crystallizing,



what are we doing anyways

- Great question
- When making crystalline candies, like rock candy, all we need to do is dissolve the sugar in water
- When we let it cool down, the sucrose molecules re-crystallize into bigger crystals
- that's where invert sugars come in (literally)

THE CHEMISTRY OF CANDY

CRYSTALLINE CANDY



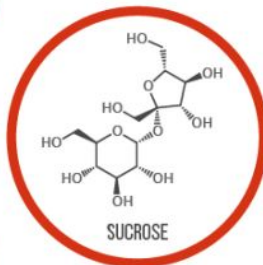
FUDGE



ROCK CANDY



FONDANT



LOWER SUGAR CONCENTRATION THAN NON-CRYSTALLINE



SUCROSE SOLUTION BOILED AT LOWER TEMPERATURE



CONTAIN MANY SMALL, FINE CRYSTALS OF SUCROSE

Generally smooth and creamy. Crystalline candies contain crystals of sucrose in their finished form; the sucrose molecules are able to align and form large lattices. They are best formed by slow cooling and little mixing of a solution for crunchy candies, and faster cooling and lots of mixing for smooth candies.

NON-CRYSTALLINE CANDY



LOLLIPOPS

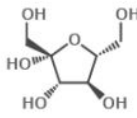
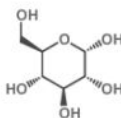


CANDY CANES



CARAMEL

INTERFERING AGENTS



HIGHER SUGAR CONCENTRATION THAN CRYSTALLINE



SUCROSE SOLUTION BOILED AT HIGHER TEMPERATURE



FROM VERY SATURATED SOLUTION - NO CRYSTALS

Generally hard & brittle. Non-crystalline, or amorphous candies, form when crystallization is prevented. This can be accomplished by the addition of sugars such as glucose and fructose that interfere with the development of crystals. Often, their mixtures are too viscous for crystals to form.



caramelization products

caramelans

$C_{24}H_{36}O_{18}$
brown color

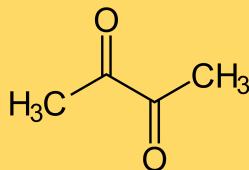
carmelens

$C_{36}H_{50}O_{25}$
brown color

caramelins

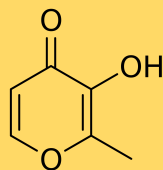
$C_{125}H_{188}O_{25}$
brown color

diacetyl



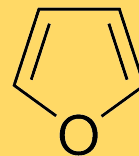
buttery

maltol

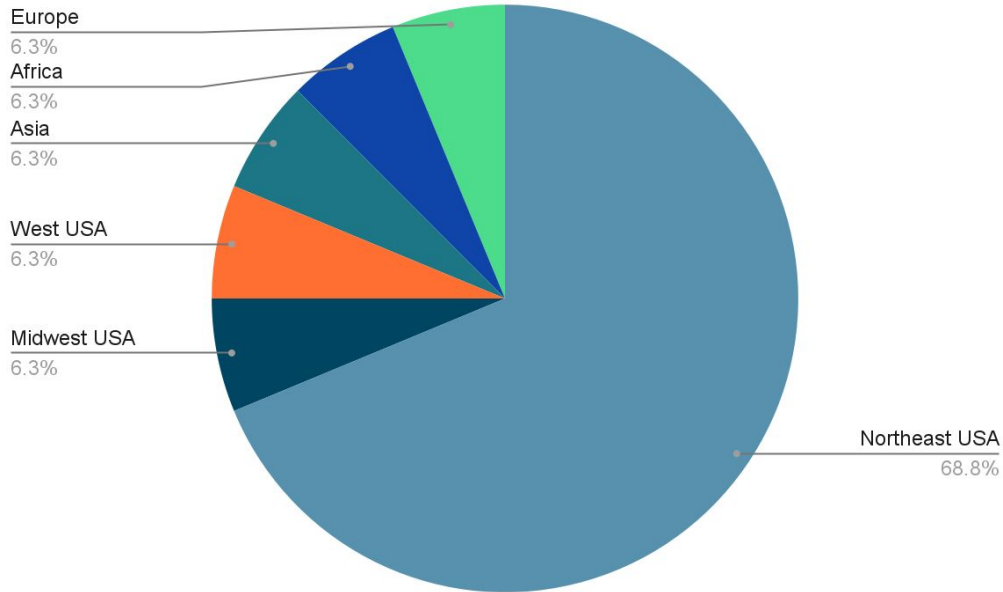


basically the
caramel taste

furan

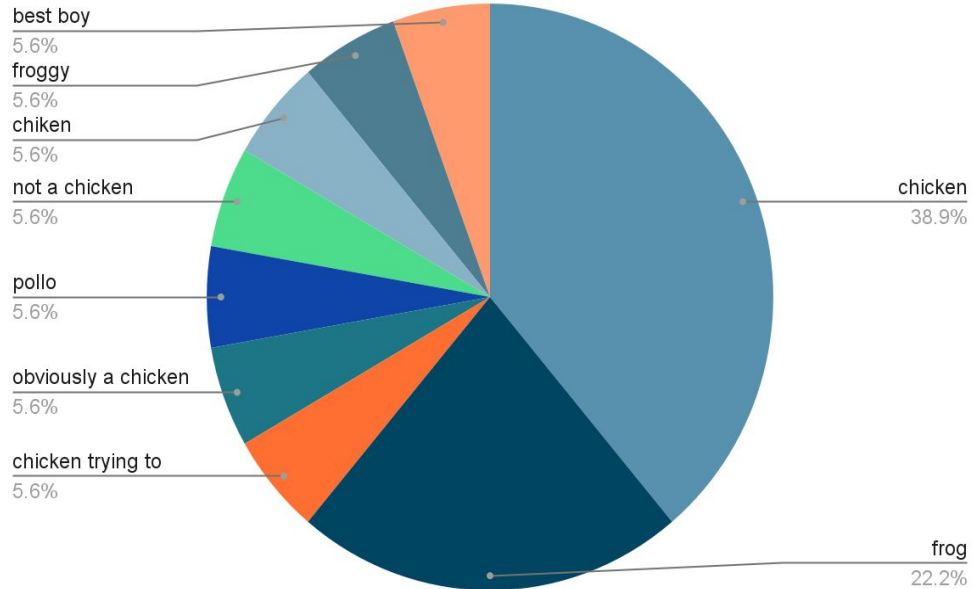


the nutty taste



**Here's where y'all
are from
(according to the
survey results)**

**Here's where y'all
thought this was
(it is a mountain
chicken)**





thanks!

Yup that's pretty much it, have a great time

After three hours, don't forget, you can cut
the caramel up however you want, and
shape it (if you want to be fancy).

alright peace out folks